

Francesco Corona (¬_¬)

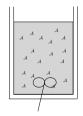
Chemical and Metallurgical Engineering School of Chemical Engineering

CHEM-LV03

 $A \to \emptyset$ and $\emptyset \to A$

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Simulation

We extend our analysis to consider a system with a single species and two reactions



We consider the degradation of some chemical species \mathcal{A} to some uninteresting form \emptyset

$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$

We also consider the production of the same species \mathcal{A} from an uninteresting form \emptyset

$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

The degradation reaction does not state that A is degraded into nothing, but that it is degraded into unmoddeled species, or that there is an outflux to another compartment

The production reaction does not state that A is created from nothing, but that it is produced from unmodelled species, or that there is an influx from another compartment

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 $\begin{array}{ccc} \mathcal{A} & \to \emptyset \text{ and } \\ \emptyset & \to \mathcal{A} \end{array}$

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$

Stochastic simulation

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 $A \to \emptyset$ and $\emptyset \to A$

 $A \to \emptyset$ and $\emptyset \to A \mid$ Simulation (cont.)



The reactions occur in a vessel of volume V

- The content of the vessel is well mixed
- → The system is in thermal equilibrium

The rate constant κ_1 is defined in such a way that the quantity (κ_1 dt) corresponds to the probability that a molecule of A is degraded in the infinitesimally small interval

$$[t, t + \mathrm{d}t)$$

The rate constant κ_2 is defined in such a way that quantity $(V \kappa_2 dt)$ is the probability that a molecule of A is produced, in the infinitesimal interval and in the unit volume

$$[t, t + dt]$$
 and V

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ightarrow \ \mathcal{A} \end{array}$

Chemical master equation Statistics

$A \to \emptyset$ and $\emptyset \to A \mid$ Simulation (cont.)

Rate constants κ_1 for degradation and κ_2 for production have different physical units

- $\rightsquigarrow \kappa_1$ is expressed in $[\sec^{-1}]$
- $\sim \kappa_2$ is expressed in [sec⁻¹m⁻³]

The probability of the degradation reaction depends on the state of the system

→ (It grows with the number of molecules available)

The probability of the production reaction depends on the size of the system

→ (It is constant, but proportional to volume)

To understand the reasoning behind the scaling of κ_2 by volume, think of dividing the container in two equally large parts, the production rate in each is also divided by two

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 $\begin{array}{ccc} \mathcal{A} \to \emptyset \text{ and } \\ \emptyset \to \mathcal{A} \end{array}$

Chemical master equation Statistics Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Simulation (cont.)

We can start our reasoning about the system in terms of time until the next reaction

$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$
$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

The state changes EITHER when one degradation OR one production reaction occurs

The probability that an event occurs in [t, t+dt) equals the probability $N_{\mathcal{A}}(t)\kappa_1 dt$ that the first reaction occurs PLUS the probability $V_{\kappa_2} dt$ that the second one occurs

$$\mathbb{P}(\text{one degradation}) = \underbrace{N_{\mathcal{A}}(t) \times \kappa_1}_{\text{N}} dt$$

$$\mathbb{P}(\text{one production}) = \underbrace{V \kappa_2}_{\text{t}} dt$$

Let $(\alpha(t) dt)$ be the probability that EITHER the first OR the second reaction occurs

$$\alpha(t) = \underbrace{N_{\mathcal{A}}(t)\kappa_1 + V\kappa_2}_{\text{Probability per unit time}}$$

We think of $\alpha(t)$ as the propensity of the system to react at time t, given its state N_A

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ightarrow \emptyset \ \mathrm{and} \ \emptyset &
ightarrow \mathcal{A} \end{array}$

equation
Statistics
Stationary
distribution

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Simulation (cont.)

To model and simulate this system of two reactions, we consider again reaction events

For the first reaction, over the infinitesimal time dt, we had

$$\mathbb{P}(\text{one degradation}) = \mathbb{P}(\text{EITHER molecule } \mathcal{A}_1 \text{ OR } \cdots \text{ OR } \mathcal{A}_{N_A(t)} \text{ degrades}) \quad (1a)$$

$$= \mathbb{P}(A_1 \text{ degrades} \cup \dots \cup A_{N_A(t)} \text{ degrades}) \tag{1b}$$

$$= \mathbb{P}(\mathcal{A}_1 \text{ degrades}) + \dots + \mathbb{P}(\mathcal{A}_{N_A(t)} \text{ degrades}) \tag{1c}$$

$$= N_{\mathcal{A}}(t) \times \mathbb{P}(\text{a molecule of } \mathcal{A} \text{ degrades}) \tag{1d}$$

$$= N_{\mathcal{A}}(t) \times \kappa_1 \mathrm{d}t \tag{1e}$$

Again, the molecules of \mathcal{A} were assumed to be indistinguishable and act independently

For the second reaction, over the infinitesimal time dt, we have

$$\mathbb{P}(\text{one production}) = V \kappa_2 dt \tag{2}$$

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Chemical mast equation Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Simulation (cont.)

$$\alpha(t) = N_{\mathcal{A}}(t)\kappa_1 + V\kappa_2$$

We use this combined information to determine the time s until the next reaction event

- \rightarrow The time s is an exponentially distributed number
- (Regardless of what reaction it will be)
- (Give the state of the system at t)

That is,

$$s \sim \text{Exp}(\alpha(t))$$

Formally, time t+s is to be understood as the exit time of the system from state $N_{\mathcal{A}}(t)$ \longrightarrow The expected (the mean) exit time is the reciprocal of $\alpha(t)$

As $\alpha(t) = N_A(t)\kappa_1 + V\kappa_2$ increases/decreases with the increase/decrease in copy numbers (other things being constant), also the mean exit time will change accordingly \sim A large/small copy numbers of reactants indicates frequent/rare reactions

$\begin{array}{ccc} \mathcal{A} & \rightarrow \ \emptyset \ \ \mathrm{and} \\ \emptyset & \rightarrow \ \mathcal{A} \end{array}$

Chemical maste equation Statistics Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Simulation (cont.)

The next reaction occurs at some time t+s, we need to determine which one it will be

The index of the next reaction is a discrete random variable

Based on their relative probability of occurrence, we have

→ For the first reaction

$$\mathbb{P}(\text{degradation occurs}) = \frac{N_{\mathcal{A}}(t)\kappa_1}{\alpha(t)}$$

 \leadsto For the second reaction

$$\mathbb{P}(\text{production occurs}) = \frac{\kappa_2 V}{\alpha(t)}$$

The next reaction type is a random variable M with a generalised Bernoulli distribution

- (A variable with multinomial distribution with two events, and one trial)
- (If one trial and whatever the number of events, a binomial distribution)

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Chemical master equation Statistics Stationary

Algorithm 1 Degradation + production | Stochastic simulation algorithm

- 1: procedure Degradation + production | SSA, exponential + multinomial Input: $n_{\mathcal{A}}(t=0) = n_{\mathcal{A}}(0)$, V, κ_1 and κ_2 Output: $(n_{\mathcal{A}}(t+s_z))_{z=1,2,...}$
- 2: Set t = 0, r = 0 and $s_z = 0$
- 3: Compute $\alpha_1(t) = n_{\mathcal{A}}(t+s_z)\kappa_1$, $\alpha_2(t+s_z) = V\kappa_2$, and $\alpha(t) = \sum \alpha_{n_r}(t)$
- 4: Compute $p = (p_1, p_2)$, with $p_{n_r} = \alpha_{n_r}/\alpha(t)$ for $n_r = 1, 2$
- 5: Compute the time s_{z+1} until next reaction

$$s_{z+1} \sim \operatorname{Exp}\left(\alpha(t)\right)$$

6: Compute the type m_{z+1} of next reaction

$$m_{z+1} \sim \text{Mult}(p, n = 1)$$

7: Set

$$n_A(t+s_{z+1}) = \begin{cases} n_A(t+s_z) + 1, & \text{if } m = 1\\ \\ n_A(t+s_z) - 1, & \text{if } m = 0 \end{cases}$$

- 8: Set $z \rightsquigarrow z + 1$
- 9: Set $t \rightsquigarrow t + s_{z+1}$
- 10: Repeat
- 11: end procedure

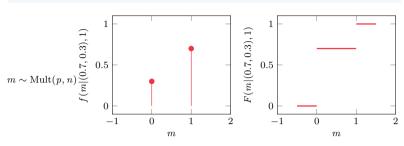
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equation Statistics Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Simulation (cont.)

Definition (Multinomial/Binomial distribution)



The probability mass function

$$f\left(m\mid p,n\right) = \begin{cases} \frac{n!}{\prod x_k!} \prod p_k^{x_k}, & \sum x_k = 1\\ 0, & \text{elsewhere} \end{cases}$$

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$A \to \emptyset$ and $\emptyset \to A$

Statistics Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Simulation (cont.)

Example

The evolution of $n_{\mathcal{A}}(t)$ is obtained by stochastic simulation for given parameter values

$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$
$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

Kinetic parameters

$$\kappa_1 = 0.1 \text{sec}^{-1}$$

$$\kappa_2 V = 1.0 \text{sec}^{-1}$$

Initial conditions

$$n_{A}(0) = 0$$

Each time the simulation is repeated, a different realisation of $(N_A(t))_{t>0}$ is obtained

$$\{n_{\mathcal{A}}^{(r)}(t_k)\}_{k=1}^K \qquad (r=1,2,\ldots,R)$$

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$

Chemical master equation

Statistics Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$

Chemical master equation

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 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Master equation (cont.)

$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$

$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

There are only three ways that lead to have $n_{\mathcal{A}}(t + \Delta t)$ molecules of \mathcal{A} , given $n_{\mathcal{A}}(t)$ $\rightarrow n_{\mathcal{A}}(t + \Delta t)$ was $n_{\mathcal{A}}(t) + 1$ and one degradation occurred in $[t, t + \Delta t)$

$$(n_A + 1)\kappa_1 \Delta t$$

 \rightarrow $n_{\mathcal{A}}(t+\Delta t)$ was $n_{\mathcal{A}}(t)-1$ and one production occurred in $[t,t+\Delta t)$

$$V {\color{red}\kappa_2} \Delta t$$

 $\rightarrow n_{\mathcal{A}}(t + \Delta t)$ was $n_{\mathcal{A}}(t)$ and no reactions occurred in $[t, t + \Delta t)$

$$1 - \left[(n_{\mathcal{A}} + 1) \kappa_1 \Delta t + V \kappa_2 \Delta t \right]$$

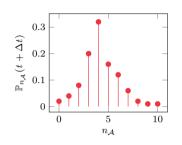
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Chemical master equation Statistics Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation

Let $\mathbb{P}_{n_A}(t+\Delta t)$ be the probability that there are n_A molecules of A at time $(t+\Delta t)$



$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$
$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

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 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation

Stationary distribution

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation

$$\mathbb{P}_{n_{\mathcal{A}}}(t + \Delta t) = \mathbb{P}_{n_{\mathcal{A}}(t)}(t) \times \underbrace{\left[1 - (n_{\mathcal{A}} + 1)\kappa_{1}\Delta t - V\kappa_{2}\Delta t\right]}_{\text{No reactions occur}} + \mathbb{P}_{n_{\mathcal{A}}(t)+1}(t) \times \underbrace{(n_{\mathcal{A}} + 1)\kappa_{1}\Delta t}_{\text{Degradation occurs}} + \mathbb{P}_{n_{\mathcal{A}}(t)-1}(t) \times \underbrace{V\kappa_{2}\Delta t}_{\text{Production occurs}}$$

After rearranging and for $\Delta t \to 0$, we get the ordinary differential difference equation

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \frac{\kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)}{\kappa_{1}}$$

 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{A}}(t)}{\mathrm{d}t} = \underbrace{\kappa_{1}(n_{A}+1)\mathbb{P}_{n_{A}+1}(t)}_{\mathrm{Gain}} - \underbrace{\kappa_{1}n_{A}\mathbb{P}_{n_{A}}(t)}_{\mathrm{Loss}} + \underbrace{\kappa_{2}V\mathbb{P}_{n_{A}-1}(t)}_{\mathrm{Gain}} - \underbrace{\kappa_{2}V\mathbb{P}_{n_{A}}(t)}_{\mathrm{Loss}}$$

$$= \kappa_{1}\left[(n_{A}+1)\mathbb{P}_{n_{A}+1}(t) - n_{A}\mathbb{P}_{n_{A}}(t)\right] + \kappa_{2}V\left[\mathbb{P}_{n_{A}-1}(t) - \mathbb{P}_{n_{A}}(t)\right]$$

Each reaction contributes to the rate of change of \mathbb{P}_{n_A} with one gain and one loss term

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \underbrace{\frac{\kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t)}{\mathrm{Degradation}} + \underbrace{\frac{\kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t)}{\mathrm{Production}}}_{\text{Gain}} - \underbrace{\frac{\kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{Degradation}}}_{\text{Loss}} - \underbrace{\frac{\kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t)}{\mathrm{Coss}}}_{\text{Loss}}$$

$$= \underbrace{\left[\kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t)\right]}_{\text{Coss}} - \underbrace{\left[\kappa_{1}n_{\mathcal{A}} + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t)\right]}_{\text{Coss}} - \underbrace{\left[\kappa_{1}n_{\mathcal{A}} + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)\right]}_{\text{Coss}} - \underbrace{\left[\kappa_{1}n_{\mathcal{A}} + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)\right]}_{\text{Coss}}}_{\text{Coss}} - \underbrace{\left[\kappa_{1}n_{\mathcal{A}} + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}$$

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 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation Statistics $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$

We know that chemical master equations (CME) are defined for all $n_A \in \{0, 1, \dots, \infty\}$

• In a degradation + production systems, there is no maximum number n_A

$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$
$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

This master equation consists of a set of infinitely many ordinary differential equations \longrightarrow The initial condition is the set $\{\mathbb{P}_{n_A}(t=0)\}_{n_A=0}^{\infty}$ of initial probabilities

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 $\mathcal{A} \to \emptyset$ and

Chemical master equation

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1} \left[(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) \right] + \kappa_{2} V \left[\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \mathbb{P}_{n_{\mathcal{A}}}(t) \right]$$
$$= \left[\kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) + \kappa_{2} V \mathbb{P}_{n_{\mathcal{A}} - 1}(t) \right] - \left[\kappa_{1} n_{\mathcal{A}} + \kappa_{2} V \mathbb{P}_{n_{\mathcal{A}}}(t) \right]$$

For each copy-number n_A and at each time t, the rate at which the probability of that number changes depends on an outgoing flow and on an incoming flow of probability

Whenever gains and losses are equal, the rate of change in probability is equal to zero

• We say that the system, from the viewpoint of $\mathbb{P}_{n_A}(t)$, is at steady-state

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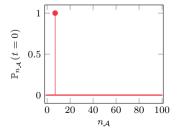
 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \frac{\kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)}{(\text{with } n_{\mathcal{A}}=0,1,\ldots,\infty)}$$

Assume to known the initial number of molecule of A and use it as initial condition



$$\mathbb{P}_{n_{\mathcal{A}}}(t=0) = \begin{cases} 1, & \text{for } n_{\mathcal{A}} = \overline{n_{\mathcal{A}}} \\ 0, & \text{elsewhere} \end{cases}$$

 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation Statistics

$\mathcal{A} o \emptyset$ and $\emptyset o \mathcal{A} \mid$ Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$
(with $n_{\mathcal{A}} = 0, 1, \dots, \infty$)

We can inspect the component equations for a first two values of the copy number n_A

For $n_{A} = 0$, we have

$$\begin{split} \frac{\mathrm{d}\mathbb{P}_0(t)}{\mathrm{d}t} &= \kappa_1(0+1)\mathbb{P}_{0+1}(t) - \kappa_1(0)\mathbb{P}_0(t) + \kappa_2 V\mathbb{P}_{0-1}(t) - \kappa_2 V\mathbb{P}_0(t) \\ &= \kappa_1\mathbb{P}_1(t) - \kappa_2 V\mathbb{P}_0(t) \end{split}$$

For $n_{A} = 1$, we have

$$\begin{split} \frac{\mathrm{d}\mathbb{P}_{1}(t)}{\mathrm{d}t} &= \kappa_{1}(1+1)\mathbb{P}_{1+1}(t) - \kappa_{1}(1)\mathbb{P}_{1}(t) + \kappa_{2} V \mathbb{P}_{1-1}(t) - \kappa_{2} V \mathbb{P}_{1}(t) \\ &= 2\kappa_{1}\mathbb{P}_{2}(t) - \kappa_{1}\mathbb{P}_{1}(t) + \kappa_{2} V \mathbb{P}_{0}(t) - \kappa_{2} V \mathbb{P}_{1}(t) \\ &= 2\kappa_{1}\mathbb{P}_{2}(t) - (\kappa_{1} + \kappa_{2} V)\mathbb{P}_{1}(t) + \kappa_{2} V \mathbb{P}_{0}(t) \end{split}$$

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 $A \to \emptyset$ and $\emptyset \to A$

Chemical maste equation Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$
(with $n_{\mathcal{A}} = 0, 1, \dots, \infty$)

To solve this master equation in practice, a truncation at some $n_{\mathcal{A}} \gg \overline{n_{\mathcal{A}}}$ can be used \leadsto The approximation is acceptable, because $\mathbb{P}_{n_{\mathcal{A}}} \to 0$ as $n_{\mathcal{A}} \to \infty$, whatever t

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Chemical master equation Statistics $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$
(with $n_{\mathcal{A}} = 0, 1, \dots, \infty$)

We can inspect the component equations for a last two values of the copy numbers n_A For $n_A = \overline{n_A}$, we have

$$\begin{split} \frac{\mathrm{d}\mathbb{P}_{\overline{n_{\mathcal{A}}}}(t)}{\mathrm{d}t} &= \kappa_{1}(\overline{n_{\mathcal{A}}}+1)\mathbb{P}_{\overline{n_{\mathcal{A}}}+1}(t) - \kappa_{1}(\overline{n_{\mathcal{A}}})\mathbb{P}_{\overline{n_{\mathcal{A}}}}(t) + \kappa_{2}\,V\mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) - \kappa_{2}\,V\mathbb{P}_{\overline{n_{\mathcal{A}}}}(t) \\ &= \kappa_{2}\,V\mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) - (\kappa_{1}\overline{n_{\mathcal{A}}} + \kappa_{2}\,V)\,\mathbb{P}_{\overline{n_{\mathcal{A}}}}(t) \end{split}$$

For $n_{\mathcal{A}} = \overline{n_{\mathcal{A}}} - 1$, we have

$$\begin{split} \frac{\mathrm{d}\mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t)}{\mathrm{d}t} &= \kappa_{1}(\overline{n_{\mathcal{A}}}-1+1)\mathbb{P}_{\overline{n_{\mathcal{A}}}-1+1}(t) - \kappa_{1}(\overline{n_{\mathcal{A}}}-1)\mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) \\ &+ \kappa_{2} V \mathbb{P}_{\overline{n_{\mathcal{A}}}-1-1}(t) - \kappa_{2} V \mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) \\ &= \kappa_{1} \overline{n_{\mathcal{A}}} \mathbb{P}_{\overline{n_{\mathcal{A}}}}(t) - \kappa_{1}(\overline{n_{\mathcal{A}}}-1)\mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) + \kappa_{2} V \mathbb{P}_{\overline{n_{\mathcal{A}}}-2}(t) - \kappa_{2} V \mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) \\ &= \kappa_{1} \overline{n_{\mathcal{A}}} \mathbb{P}_{\overline{n_{\mathcal{A}}}}(t) - \left[\kappa_{1}(\overline{n_{\mathcal{A}}}-1) + \kappa_{2} V\right] \mathbb{P}_{\overline{n_{\mathcal{A}}}-1}(t) + \kappa_{2} V \mathbb{P}_{\overline{n_{\mathcal{A}}}-2}(t) \end{split}$$

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 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation

Stationary

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Master equation (cont.)

example

The solution to the chemical master equation

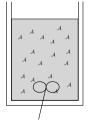
$$\mathcal{A} \xrightarrow{\kappa_1} \emptyset$$
$$\emptyset \xrightarrow{\kappa_2} \mathcal{A}$$

Kinetic parameters and initial conditions

$$\kappa_1 = 0.1 \text{sec}^{-1}$$

$$\kappa_2 V = 1.0 \text{sec}^{-1}$$

$$n_{\mathcal{A}}(0) = 0$$



 $A \to \emptyset$ and $\emptyset \to A$

Chemical master

Statistics

Stationary

$$\mathcal{A} o \emptyset$$
 and $\emptyset o \mathcal{A}$

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 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation
Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

In addition, we have derived the chemical master equation for all values $n_{\mathcal{A}}=0,1,\ldots,\infty$

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \frac{\kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)}{\kappa_{1}}$$

That is,

$$\begin{split} \frac{\mathrm{d}\mathbb{P}_0(t)}{\mathrm{d}t} &= \kappa_1 \mathbb{P}_1(t) - \kappa_2 \, V \mathbb{P}_0(t) \\ \frac{\mathrm{d}\mathbb{P}_1(t)}{\mathrm{d}t} &= 2\kappa_1 \mathbb{P}_2(t) - (\kappa_1 + \kappa_2 \, V) \, \mathbb{P}_1(t) + \kappa_2 \, V \mathbb{P}_0(t) \\ \frac{\mathrm{d}\mathbb{P}_2(t)}{\mathrm{d}t} &= 3\kappa_1 \mathbb{P}_3(t) - (2\kappa_1 + \kappa_2 \, V) \, \mathbb{P}_2(t) + \kappa_2 \, V \mathbb{P}_1(t) \\ & \cdots = \cdots \\ \frac{\mathrm{d}\mathbb{P}_{\overline{n_A} - 1}(t)}{\mathrm{d}t} &= \kappa_1 \, \overline{n_A} \mathbb{P}_{\overline{n_A}}(t) - \left[\kappa_1(\overline{n_A} - 1) + \kappa_2 \, V\right] \mathbb{P}_{\overline{n_A} - 1}(t) + \kappa_2 \, V \mathbb{P}_{\overline{n_A} - 2}(t) \\ \frac{\mathrm{d}\mathbb{P}_{\overline{n_A}}(t)}{\mathrm{d}t} &= \kappa_2 \, V \mathbb{P}_{\overline{n_A} - 1}(t) - (\kappa_1 \, \overline{n_A} + \kappa_2 \, V) \, \mathbb{P}_{\overline{n_A}}(t) \end{split}$$

CHEM-LV03

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ Chemical master equation

Statistics Stationary distribution

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics

We can use the master equation to determine the expected evolution $(\mathbb{E}[N_{\mathcal{A}}(t)])_{t\geq 0}$ In general, we have

$$\begin{split} \mathbf{E}\left[N_{\mathcal{A}}(t)\right] &= \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= M_{\mathcal{A}}(t) \end{split}$$

We can also use the master equation to determine the spread around $(\mathbb{E}[N_{\mathcal{A}}(t)])_{t\geq 0}$ In general, we have

$$E\left[\left(N_{\mathcal{A}}(t) - M_{\mathcal{A}}(t)\right)^{2}\right] = \sum_{n_{\mathcal{A}}=0}^{\infty} \left(n_{\mathcal{A}} - M_{\mathcal{A}}(t)\right)^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$
$$= V_{\mathcal{A}}(t)$$

CHEM-LV03 2022

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ Chemical mastequation
Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

By multiplying each component equation of the master equation by n_A , we obtain

$$\begin{split} n_{\mathcal{A}} \frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} &= \kappa_{1} n_{\mathcal{A}}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1} n_{\mathcal{A}}^{2}\mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{n_{\mathcal{A}}}(t) \quad \text{(for all } n_{\mathcal{A}} = 0, 1, \dots, \infty) \end{split}$$

That is,

$$\begin{split} n_{\mathcal{A}} \frac{\mathrm{d}\mathbb{P}_{0}(t)}{\mathrm{d}t} &= \kappa_{1} n_{\mathcal{A}} \mathbb{P}_{1}(t) - \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{0}(t) \\ n_{\mathcal{A}} \frac{\mathrm{d}\mathbb{P}_{1}(t)}{\mathrm{d}t} &= 2\kappa_{1} n_{\mathcal{A}} \mathbb{P}_{2}(t) - (\kappa_{1} + \kappa_{2} V) n_{\mathcal{A}} \mathbb{P}_{1}(t) + \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{0}(t) \\ n_{\mathcal{A}} \frac{\mathrm{d}\mathbb{P}_{2}(t)}{\mathrm{d}t} &= 3\kappa_{1} n_{\mathcal{A}} \mathbb{P}_{3}(t) - (2\kappa_{1} + \kappa_{2} V) n_{\mathcal{A}} \mathbb{P}_{2}(t) + \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{1}(t) \\ &\cdots = \cdots \end{split}$$

Summing over all the values of n_{Δ} , we get

$$\begin{split} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} &= \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{1} n_{\mathcal{A}}(n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}+1}(t) - \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{1} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation

Statistics Stationary $\mathcal{A} o \emptyset$ and $\emptyset o \mathcal{A}$ | Statistics (cont.)

$$\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{1} n_{\mathcal{A}}(n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}+1}(t) - \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{1} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$
$$+ \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \sum_{n_{\mathcal{A}}=0}^{\infty} \kappa_{2} n_{\mathcal{A}} V \mathbb{P}_{n_{\mathcal{A}}}(t)$$

Rearranging terms, we get

$$\frac{\mathrm{d}}{\mathrm{d}t} \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{M_{\mathcal{A}}(t)} = \underbrace{\kappa_{1}}_{n_{\mathcal{A}}=0} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}(n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

$$+ \frac{\kappa_2 V}{\sum_{n_{\mathcal{A}} = 1}^{\infty}} \sum_{n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_2 V} \underbrace{\sum_{n_{\mathcal{A}} = 0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{M_{\mathcal{A}}(t)}$$

CHEM-LV03 2022

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ Chemical maste

Statistics

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2}\mathbb{P}_{n_{\mathcal{A}}}(t)$$

$$+ \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t)$$

Changing the indexes $(n_A + 1) \rightsquigarrow n_A$ and $n_A \rightsquigarrow (n_A - 1)$ in the first term, we write

$$\begin{split} \frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} &= \kappa_{1} \sum_{n_{\mathcal{A}}=-1}^{\infty} (n_{\mathcal{A}}-1)n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &- \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}-1)n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &- \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

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 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$

Chemical master equation Statistics

Stationary distribution

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

We can write the ordinary differential equation for the evolution of the process' mean

$$\begin{split} \frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}(n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

$\begin{array}{c} \mathrm{CHEM\text{-}LV03} \\ 2022 \end{array}$

 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\begin{split} \frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} - 1) n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

Changing the indexes $(n_A - 1) \rightsquigarrow n_A$ and $n_A \rightsquigarrow (n_A + 1)$ in the third term, we write

$$\begin{split} \frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} - 1)n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}+1=1}^{\infty} (n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} - 1)n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \underbrace{\sum_{n_{\mathcal{A}}=1}^{\infty} (n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{n_{\mathcal{A}}=0} \end{split}$$

 $A \to \emptyset$ and $\emptyset \to A$

Chemical maste equation Statistics

Stationary

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = \underbrace{\kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} - 1) n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{+ \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

Combining the first and second term (gain and loss due to degradation), we get

$$\begin{split} \frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} \underbrace{\left[(n_{\mathcal{A}}-1)n_{\mathcal{A}}-n_{\mathcal{A}}^{2} \right]}_{-n_{\mathcal{A}}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= -\kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

CHEM-LV03 2022

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation

Chemical mas equation Statistics Stationary $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = -\kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

Because we have that $\sum n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) = M_{\mathcal{A}}(t)$ and $\sum \mathbb{P}_{n_{\mathcal{A}}}(t) = 1$, we have

$$\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = -\kappa_{1}M_{\mathcal{A}}(t) + \kappa_{2}V$$

The equation of motion for the expected value of $(N_{\mathcal{A}}(t))_{t>0}$

To integrate it, we need to specify the initial condition,

$$M_{\mathcal{A}}(t=0) = M_{\mathcal{A}}(0)$$

The solution can be written in closed form,

$$M_{\mathcal{A}}(t) = \frac{\kappa_2}{\kappa_1} V + \left(M_{\mathcal{A}}(0) - \frac{\kappa_2}{\kappa_1} V \right) \exp\left(-\kappa_1 t \right)$$

CHEM-LV03 2022

 $egin{array}{l} \mathcal{A} &
ightarrow \emptyset \ \mathrm{and} \ \emptyset &
ightarrow \mathcal{A} \end{array}$

Chemical maste equation Statistics

Statistics Stationary $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid \text{Statistics (cont.)}$

$$\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} - 1)n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$
$$+ \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

Combining the third and fourth terms (gain and loss from production), we get

$$\begin{split} \frac{\mathrm{d}M_{A}(t)}{\mathrm{d}t} &= -\kappa_{1} \sum_{n_{A}=0}^{\infty} n_{A} \mathbb{P}_{n_{A}}(t) \\ &+ \kappa_{2} V \sum_{n_{A}=0}^{\infty} (n_{A}+1) \mathbb{P}_{n_{A}}(t) - \kappa_{2} V \sum_{n_{A}=0}^{\infty} n_{A} \mathbb{P}_{n_{A}}(t) \\ &= -\kappa_{1} \sum_{n_{A}=0}^{\infty} n_{A} \mathbb{P}_{n_{A}}(t) + \kappa_{2} V \sum_{n_{A}=0}^{\infty} \underbrace{\left[(n_{A}+1) - n_{A} \right]}_{1} \mathbb{P}_{n_{A}}(t) \\ &= -\kappa_{1} \sum_{n_{A}=0}^{\infty} n_{A} \mathbb{P}_{n_{A}}(t) + \kappa_{2} V \sum_{n_{A}=0}^{\infty} \mathbb{P}_{n_{A}}(t) \end{split}$$

CHEM-LV03 2022

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

Example

We consider the evolution of the mean process from an initial condition $M_A(0) = 0$

$$\begin{array}{c} \mathcal{A} \xrightarrow{\kappa_1} \emptyset \\ \emptyset \xrightarrow{\kappa_2} \mathcal{A} \end{array}$$

Kinetic parameters and initial conditions

$$\kappa_1 = 0.1 \text{sec}^{-1}$$

$$\kappa_2 V = 1.0 \text{sec}^{-1}$$

$$n_{\mathcal{A}}(0) = 0$$



We have,

$$\begin{split} M_{\mathcal{A}}(t) &= \frac{\kappa_2}{\kappa_1} V + (M_{\mathcal{A}}(0) - \frac{\kappa_2}{\kappa_1} V) \exp\left(-\kappa_1 t\right) \\ &= \frac{\kappa_2}{\kappa_1} V - \frac{\kappa_2}{\kappa_1} V \exp\left(-\kappa_1 t\right) \\ &= \frac{\kappa_2}{\kappa_1} V \left(1 - \exp\left(-\kappa_1 t\right)\right) \end{split}$$

Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Statistics (cont.)

We can derive an ordinary differential equation for the evolution of the process' variance We start by considering the definition of variance of the process at time t

$$\begin{split} &\mathbf{E}\left[\left(N_{\mathcal{A}}(t)-M_{\mathcal{A}}(t)\right)^{2}\right] = \sum_{n_{\mathcal{A}}=0}^{\infty}\left[n_{\mathcal{A}}-M_{\mathcal{A}}(t)\right]^{2}\mathbb{P}_{n_{\mathcal{A}}}(t) \\ &=\sum_{n_{\mathcal{A}}=0}^{\infty}\left[n_{\mathcal{A}}^{2}-2n_{\mathcal{A}}M_{\mathcal{A}}(t)+M_{\mathcal{A}}(t)^{2}\right]\mathbb{P}_{n_{\mathcal{A}}}(t) \\ &=\sum_{n_{\mathcal{A}}=0}^{\infty}n_{\mathcal{A}}^{2}\mathbb{P}_{n_{\mathcal{A}}}(t)-2M_{\mathcal{A}}(t)\underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t)}_{M_{\mathcal{A}}(t)} \\ &+M_{\mathcal{A}}(t)^{2}\underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty}\mathbb{P}_{n_{\mathcal{A}}}(t)}_{=1} \\ &=\sum_{n_{\mathcal{A}}=0}^{\infty}n_{\mathcal{A}}^{2}\mathbb{P}_{n_{\mathcal{A}}}(t)-M_{\mathcal{A}}(t)^{2} \\ &=V_{\mathcal{A}}(t) \end{split}$$

CHEM-LV03 2022

Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Statistics (cont.)

In addition, for all values $n_{\mathcal{A}} = 0, 1, \dots, \infty$, we derived the chemical master equation

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$

By multiplying each component $(n_A = 0, 1, ..., \infty)$ of the master equation by n_A^2

$$n_{\mathcal{A}}^{2} \frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1} n_{\mathcal{A}}^{2} (n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1} n_{\mathcal{A}}^{3} \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} n_{\mathcal{A}}^{2} V \mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2} n_{\mathcal{A}}^{2} V \mathbb{P}_{n_{\mathcal{A}}}(t)$$

After summing over n_A and rearranging the terms, we get

$$\frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) = \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 (n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^3 \mathbb{P}_{n_{\mathcal{A}}}(t)$$

$$+ \kappa_2 V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_2 V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t)$$

CHEM-LV03 2022

Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid \mathbf{Statistics}$ (cont.)

$$\sum_{n_{\mathcal{A}}=0}^{\infty}n_{\mathcal{A}}{}^{2}\mathbb{P}_{n_{\mathcal{A}}}(t)=V_{\mathcal{A}}(t)+M_{\mathcal{A}}(t)^{2}$$

That is, we have quickly found a relationship between the process' mean and variance (This relation is general, not system specific, it will be used later on)

By taking the derivative of both sides with respect to time, we get

$$\frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) = \frac{\mathrm{d}}{\mathrm{d}t} \left[V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^{2} \right]$$

$$= \frac{\mathrm{d}}{\mathrm{d}t} V_{\mathcal{A}}(t) + \frac{\mathrm{d}}{\mathrm{d}t} M_{\mathcal{A}}(t)^{2}$$

$$= \frac{\mathrm{d}}{\mathrm{d}t} V_{\mathcal{A}}(t) + 2M_{\mathcal{A}}(t)$$

Rearranging, we have

$$\frac{\mathrm{d}V_{\mathcal{A}}(t)}{\mathrm{d}t} = \underbrace{\frac{\mathrm{d}}{\mathrm{d}t} \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{!} - 2 \underbrace{M_{\mathcal{A}}(t)}_{!}$$

CHEM-LV03

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Statistics (cont.)

$$\frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) = \kappa_{1} \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} (n_{\mathcal{A}}+1) \mathbb{P}_{n_{\mathcal{A}}+1}(t)}_{+\kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

Changing indexes $(n_A + 1) \rightsquigarrow n_A$ and $n_A \rightsquigarrow (n_A - 1)$ in the first sum, we get

$$\frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) = \kappa_{1} \underbrace{\sum_{n_{\mathcal{A}}=1=0}^{\infty} (n_{\mathcal{A}}-1)^{2} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{n_{\mathcal{A}}=1} - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{3} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

$$+ \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

$$= \kappa_{1} \underbrace{\sum_{n_{\mathcal{A}}=1}^{\infty} (n_{\mathcal{A}}-1)^{2} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{3} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{n_{\mathcal{A}}=0}$$

$$+ \kappa_{2} V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation

Statistics Stationary $\mathcal{A} o \emptyset$ and $\emptyset o \mathcal{A}$ | Statistics (cont.)

$$\begin{split} \frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) &= \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}-1)^2 n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^3 \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_2 V \sum_{n_{\mathcal{A}}=1}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_2 V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

Changing indexes $(n_A-1) \leadsto n_A$ and $n_A \leadsto (n_A+1)$ in the third sum, we get

$$\begin{split} \frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}-1)^{2} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{3} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}+1=1}^{\infty} (n_{\mathcal{A}}+1)^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}-1)^{2} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{3} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}+1)^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

CHEM-LV03 2022

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation

Statistics

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\begin{split} \frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) &= \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} (-2n_{\mathcal{A}}^2 + n_{\mathcal{A}}) \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_2 V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}} + 1)^2 \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_2 V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

Combining the third and fourth sum, we get

$$\begin{split} \frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} \left(-2n_{\mathcal{A}}^{2} + n_{\mathcal{A}}\right) \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} \underbrace{\left[(n_{\mathcal{A}}+1)^{2} - n_{\mathcal{A}}^{2}\right]}_{2n_{\mathcal{A}}+1} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} \left(-2n_{\mathcal{A}}^{2} + n_{\mathcal{A}}\right) \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} \left(2n_{\mathcal{A}} + 1\right) \mathbb{P}_{n_{\mathcal{A}}}(t) T \end{split}$$

CHEM-LV03 2022 $\mathcal{A} o \emptyset \,\, ext{and} \,\, \emptyset o \mathcal{A} \,\, | \,\, ext{Statistics (cont.)}$

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ Chemical master equation

Statistics

$$\frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) = \underbrace{\kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}-1)^{2} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{3} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{+ \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}+1)^{2} \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)$$

Combining the first and second sum, we get

$$\begin{split} \frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) &= \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} \underbrace{\left[(n_{\mathcal{A}}-1)^2 n_{\mathcal{A}} - n_{\mathcal{A}}^3 \right]}_{-2n_{\mathcal{A}}^2 + n_{\mathcal{A}}} \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_2 \, V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}+1)^2 \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_2 \, V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} \left(-2n_{\mathcal{A}}^2 + n_{\mathcal{A}} \right) \mathbb{P}_{n_{\mathcal{A}}}(t) \\ &+ \kappa_2 \, V \sum_{n_{\mathcal{A}}=0}^{\infty} (n_{\mathcal{A}}+1)^2 \mathbb{P}_{n_{\mathcal{A}}}(t) - \kappa_2 \, V \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) \end{split}$$

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 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation
Statistics

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\frac{\mathrm{d}}{\mathrm{d}t} \sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^2 \mathbb{P}_{n_{\mathcal{A}}}(t) = \kappa_1 \sum_{n_{\mathcal{A}}=0}^{\infty} \left(-2n_{\mathcal{A}}^2 + n_{\mathcal{A}}\right) \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_2 V \sum_{n_{\mathcal{A}}=0}^{\infty} \left(2n_{\mathcal{A}} + 1\right) \mathbb{P}_{n_{\mathcal{A}}}(t)$$

We can use some of the identities introduced earlier to get

$$\frac{\mathrm{d}}{\mathrm{d}t} \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^{2}} = \kappa_{1} \sum_{n_{\mathcal{A}}=0}^{\infty} (-2n_{\mathcal{A}}^{2} + n_{\mathcal{A}}) \mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2} V \sum_{n_{\mathcal{A}}=0}^{\infty} (2n_{\mathcal{A}} + 1) \mathbb{P}_{n_{\mathcal{A}}}(t)$$

$$= -2\kappa_{1} \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}}^{2} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^{2}} + \kappa_{1} \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} n_{\mathcal{A}} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{M_{\mathcal{A}}(t)} + \kappa_{2} V \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{n_{\mathcal{A}}=0} + \kappa_{2} V \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{M_{\mathcal{A}}(t)} + \kappa_{2} V \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} \mathbb{P}_{n_{\mathcal{A}}}(t)}_{n_{\mathcal{A}}=0} + \kappa_{2} V \underbrace{\sum_{n_{\mathcal{A}}=0}^{\infty} \mathbb{P$$

 $egin{pmatrix} \mathcal{A} & \to \ \emptyset \ \ \mathrm{and} \ \emptyset & \to \ \mathcal{A} \ \end{matrix}$

Chemical master equation

Stationary

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\frac{\mathrm{d}}{\mathrm{d}t} \left[V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^2 \right] = -2\kappa_1 \left[V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^2 \right] + \kappa_1 M_{\mathcal{A}}(t) + 2\kappa_2 V M_{\mathcal{A}}(t) + \kappa_2 V$$

Taking the derivative with respect to time, we obtain

$$\frac{\mathrm{d}V_{\mathcal{A}}(t)}{\mathrm{d}t} + 2M_{\mathcal{A}}(t)\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = -2\kappa_1 \left[V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^2\right] + \kappa_1 M_{\mathcal{A}}(t) + 2\kappa_2 V M_{\mathcal{A}}(t) + \kappa_2 V$$

Rearranging, we get

$$\begin{split} \frac{\mathrm{d}V_{\mathcal{A}}(t)}{\mathrm{d}t} &= -2M_{\mathcal{A}}(t) \underbrace{\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t}}_{-\kappa_1 M_{\mathcal{A}}(t) + \kappa_2 V} -2\kappa_1 \left[V_{\mathcal{A}}(t) + M_{\mathcal{A}}(t)^2 \right] \\ &+ \kappa_1 M_{\mathcal{A}}(t) + 2\kappa_2 V M_{\mathcal{A}}(t) + \kappa_2 V \\ &= -2\kappa_1 V_{\mathcal{A}}(t) + \kappa_1 M_{\mathcal{A}}(t) + \kappa_2 V \end{split}$$

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 $A \to \emptyset$ and $\emptyset \to A$ Chemical master equation

Statistics Stationary $\mathcal{A} o \emptyset$ and $\emptyset o \mathcal{A} \mid$ Statistics (cont.)

$$\begin{split} \frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} &= -\kappa_1 M_{\mathcal{A}}(t) + \kappa_2 V \\ \frac{\mathrm{d}V_{\mathcal{A}}(t)}{\mathrm{d}t} &= -2\kappa_1 V_{\mathcal{A}}(t) + \kappa_1 M_{\mathcal{A}}(t) + \kappa_2 V \end{split}$$

We are interested in the expected value and spread of the process after an infinite time $\frac{1}{2}$

Both the expected value and the spread of the process approach steady-state

$$M_{\mathcal{A}}^{\text{ss}} = \lim_{t \to \infty} M_{\mathcal{A}}(t)$$
$$V_{\mathcal{A}}^{\text{ss}} = \lim_{t \to \infty} V_{\mathcal{A}}(t)$$

By letting the derivatives to be equal to zero, we get

$$0 = -\kappa_1 M_A^{\text{ss}} + \kappa_2 V$$

$$0 = -2\kappa_1 V_A^{\text{ss}} + \kappa_1 M_A^{\text{ss}} + \kappa_2 V$$

After solving the set of equations, we get

$$M_{\mathcal{A}}^{\mathrm{ss}} = V_{\mathcal{A}}^{\mathrm{ss}} = \frac{\kappa_2}{\kappa_1} V$$

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 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation Statistics

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Statistics (cont.)

$$\frac{\mathrm{d}M_{\mathcal{A}}(t)}{\mathrm{d}t} = -\kappa_1 M_{\mathcal{A}}(t) + \kappa_2 V$$

$$\frac{\mathrm{d}V_{\mathcal{A}}(t)}{\mathrm{d}t} = -2\kappa_1 V_{\mathcal{A}}(t) + \kappa_1 M_{\mathcal{A}}(t) + \kappa_2 V$$

The equations of motion for the mean and variance process can be integrated in time

- Initial conditions must be provided
- That is, $M_A(0)$ and $V_A(0)$

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 $A \to \emptyset$ and $\emptyset \to A$

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Stationary distribution

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$

Stationary distribution

 $A \to \emptyset$ and $\emptyset \to A$

Chemical master equation

Stationary distribution

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Stationary distribution

It is interesting to understand what is the limit probability distribution $\mathbb{P}_{n_A}(t \to \infty)$

$$\pi_{n_{\mathcal{A}}} = \lim_{t \to \infty} \mathbb{P}_{n_{\mathcal{A}}}(t)$$
 (for all $n_{\mathcal{A}} = 0, 1, \dots$)

One way to determine this limit distribution is by running many stochastic simulations for a sufficiently long time, then build an empirical approximation of the distribution

Alternatively, we can consider the steady-state form of the chemical master equation

• For all $n_A = 0, 1, \ldots$, we have

$$\underbrace{\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t}}_{0} = \kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$

CHEM-LV03

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ Chemical master equation

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A} \mid$ Stationary distribution (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}\,V\mathbb{P}_{n_{A}-1}(t) - \kappa_{2}\,V\mathbb{P}_{n_{A}}(t)$$

Similarly, for $n_A = 1$ molecules

$$\begin{split} 0 &= \kappa_1 (n_{\mathcal{A}} + 1) \pi_{n_{\mathcal{A}} + 1} - \kappa_1 n_{\mathcal{A}} \pi_{n_{\mathcal{A}}} + \kappa_2 V \pi_{n_{\mathcal{A}} - 1} - \kappa_2 V \pi_{n_{\mathcal{A}}} \\ &= \kappa_1 (2) \pi_{n_{\mathcal{A}} = 2} - \kappa_1 (1) \pi_{n_{\mathcal{A}} = 1} + \kappa_2 V \pi_{n_{\mathcal{A}} = 0} - \kappa_2 V \pi_{n_{\mathcal{A}} = 1} \\ &= 2 \kappa_1 \pi_{n_{\mathcal{A}} = 2} - \kappa_1 \pi_{n_{\mathcal{A}} = 1} + \kappa_2 V \pi_{n_{\mathcal{A}} = 0} - \kappa_2 V \pi_{n_{\mathcal{A}} = 1} \\ &= 2 \kappa_1 \pi_{n_{\mathcal{A}} = 2} - (\kappa_1 + \kappa_2 V) \pi_{n_{\mathcal{A}} = 1} + \kappa_2 V \pi_{n_{\mathcal{A}} = 0} \end{split}$$

This yields the relation between the long-term probabilities of $n_A = 0, 1$ and $n_A = 2$

$$\pi_{n_{\mathcal{A}}=2} = \frac{1}{2} \left(\frac{\kappa_1 + \kappa_2 V}{\kappa_1} \right) \pi_{n_{\mathcal{A}}=1} - \frac{1}{2} \left(\frac{\kappa_2}{\kappa_1} V \right) \pi_{n_{\mathcal{A}}=0}$$

CHEM-LV03 2022 $\mathcal{A} o \emptyset$ and $\emptyset o \mathcal{A} \mid$ Stationary distribution (cont.)

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation

Statistics

Chemical master equation Statistics Stationary distribution $\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$

For the case of $n_A = 0$ molecules, at $t = \infty$ we get

$$\begin{split} 0 &= \kappa_1(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_1 n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_2 \, V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_2 \, V\mathbb{P}_{n_{\mathcal{A}}}(t) \\ &= \kappa_1(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(\infty) - \kappa_1 n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(\infty) + \kappa_2 \, V\mathbb{P}_{n_{\mathcal{A}}-1}(\infty) - \kappa_2 \, V\mathbb{P}_{n_{\mathcal{A}}}(\infty) \\ &= \kappa_1(n_{\mathcal{A}}+1)\pi_{n_{\mathcal{A}}+1} - \kappa_1 n_{\mathcal{A}}\pi_{n_{\mathcal{A}}} + \kappa_2 \, V\pi_{n_{\mathcal{A}}-1} - \kappa_2 \, V\pi_{n_{\mathcal{A}}} \\ &= \kappa_1(1)\pi_{n_{\mathcal{A}}=1} - \kappa_1(0)\pi_{n_{\mathcal{A}}=0} + \kappa_2 \, V\pi_{n_{\mathcal{A}}=-1} - \kappa_2 \, V\pi_{n_{\mathcal{A}}=0} \\ &= \kappa_1\pi_{n_{\mathcal{A}}=1} - \kappa_2 \, V\pi_{n_{\mathcal{A}}=0} \end{split}$$

This yields the relationship between the long-term probabilities of $n_A = 0$ and $n_A = 1$

$$\pi_{n_{\mathcal{A}}=1} = \left(\frac{\kappa_2}{\kappa_1}V\right)\pi_{n_{\mathcal{A}}=0},$$

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 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master

Statistics Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Stationary distribution (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}}+1)\mathbb{P}_{n_{\mathcal{A}}+1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}-1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$

Then, for $n_A = 2$ molecules

$$\begin{split} 0 &= \kappa_1 (n_{\mathcal{A}} + 1) \pi_{n_{\mathcal{A}} + 1} - \kappa_1 n_{\mathcal{A}} \pi_{n_{\mathcal{A}}} + \kappa_2 V \pi_{n_{\mathcal{A}} - 1} - \kappa_2 V \pi_{n_{\mathcal{A}}} \\ &= \kappa_1 (3) \pi_{n_{\mathcal{A}} = 3} - \kappa_1 (2) \pi_{n_{\mathcal{A}} = 2} + \kappa_2 V \pi_{n_{\mathcal{A}} = 1} - \kappa_2 V \pi_{n_{\mathcal{A}} = 2} \\ &= 3 \kappa_1 \pi_{n_{\mathcal{A}} = 3} - 2 \kappa_1 \pi_{n_{\mathcal{A}} = 2} + \kappa_2 V \pi_{n_{\mathcal{A}} = 1} - \kappa_2 V \pi_{n_{\mathcal{A}} = 2} \\ &= 3 \kappa_1 \pi_{n_{\mathcal{A}} = 3} - 2 (\kappa_1 + \kappa_2 V) \pi_{n_{\mathcal{A}} = 2} + \kappa_2 V \pi_{n_{\mathcal{A}} = 1} \end{split}$$

This yields the relation between the long-term probabilities of $n_A = 1, 2$ and $n_A = 3$

$$\pi_{n_{\mathcal{A}}=3}=\frac{1}{2}\left[\frac{2(\kappa_{1}+\kappa_{2}\,V)}{\kappa_{1}}\right]\pi_{n_{\mathcal{A}}=2}-\frac{1}{3}\left(\frac{\kappa_{2}\,V}{\kappa_{1}}\right)\pi_{n_{\mathcal{A}}=1}$$

 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Stationary distribution (cont.)

$$\frac{\mathrm{d}\mathbb{P}_{n_{\mathcal{A}}}(t)}{\mathrm{d}t} = \kappa_{1}(n_{\mathcal{A}} + 1)\mathbb{P}_{n_{\mathcal{A}} + 1}(t) - \kappa_{1}n_{\mathcal{A}}\mathbb{P}_{n_{\mathcal{A}}}(t) + \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}} - 1}(t) - \kappa_{2}V\mathbb{P}_{n_{\mathcal{A}}}(t)$$

For the general case, the steady-state chemical master equation for $n_A = 1, 2, \dots$

$$0 = \kappa_1(n_A + 1)\pi_{n_A+1} - \kappa_1 n_A \pi_{n_A} + \kappa_2 V \pi_{n_A-1} - \kappa_2 V \pi_{n_A}$$

This yields the relation between the long-term probabilities of $n_A - 1$, n_A and $n_A + 1$

$$\pi_{n_{\mathcal{A}}+1} = \frac{1}{\kappa_1(n_{\mathcal{A}}+1)} \left[\left(\kappa_1 n_{\mathcal{A}} + \kappa_2 V \right) \pi_{n_{\mathcal{A}}} - \kappa_2 V \pi_{n_{\mathcal{A}}-1} \right]$$

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 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ Chemical master equation Statistics Stationary distribution

$\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Stationary distribution (cont.)

The recursion from the steady-state master equation has also a closed-form solution

$$\pi_{n_{\mathcal{A}}} = \frac{C}{n_{\mathcal{A}}!} \left(\frac{\kappa_2 V}{\kappa_1}\right)^{n_{\mathcal{A}}} \quad \text{(for all } n_{\mathcal{A}} = 0, 1, \dots \text{ and } C \in \mathbb{R}\text{)}$$

To determine constant C, substitute π_{n_A} in the normalisation constraint

$$1 = \sum_{n_{\mathcal{A}}=0}^{\infty} \frac{C}{n_{\mathcal{A}}!} \left(\frac{\kappa_2 V}{\kappa_1}\right)^{n_{\mathcal{A}}}$$
$$= C \sum_{n_{\mathcal{A}}=0}^{\infty} \frac{1}{n_{\mathcal{A}}!} \left(\frac{\kappa_2 V}{\kappa_1}\right)^{n_{\mathcal{A}}}$$
$$= C \exp\left(\frac{\kappa_2 V}{\kappa_1}\right)$$

Rearranging, we get

$$C = \exp\left(-\frac{\kappa_2 V}{\kappa_1}\right)$$

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Chemical mast equation Statistics

Stationary distribution We can iteratively obtain π_{n_A} for all values of $n_A = 1, 2, \dots, \infty$, from a known $\pi_{n_A = 0}$

 $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Stationary distribution (cont.)

$$\pi_{n_{\mathcal{A}}+1} = \frac{1}{\kappa_1(n_{\mathcal{A}}+1)} \left[(\kappa_1 n_{\mathcal{A}} + \kappa_2 V) \pi_{n_{\mathcal{A}}} - \kappa_2 V \pi_{n_{\mathcal{A}}-1} \right]$$

After computing $\{\pi_{n_A}\}_{n_A=0}^{\overline{n_A}\gg 0}$, we re-normalise to satisfy the usual closure constraint

$$\sum_{n_{\mathcal{A}}=0}^{\infty} \pi_{n_{\mathcal{A}}} = 1$$

That is, for all n_A we re-compute all the π_{n_A}

$$\pi_{n_{\mathcal{A}}} = \frac{\pi_{n_{\mathcal{A}}}}{\sum_{n_{\mathcal{A}}=0}^{\overline{n_{\mathcal{A}}}} \pi_{n_{\mathcal{A}}}} \qquad (n_{\mathcal{A}} = 1, 2, \dots, \infty)$$

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 $A \rightarrow \emptyset$ and $\emptyset \rightarrow A$ Chemical master equation

Stationary distribution $\mathcal{A} \to \emptyset$ and $\emptyset \to \mathcal{A}$ | Stationary distribution (cont.)

The resulting stationary distribution π_{n_A} of copy numbers is the Poisson distribution

$$\pi_{n_{\mathcal{A}}} = \frac{1}{n_{\mathcal{A}}!} \left(\frac{\kappa_2 V}{\kappa_1} \right)^{n_{\mathcal{A}}} \exp\left(-\frac{\kappa_2 V}{\kappa_1} \right)$$

Transient distributions are also Poisson, with mean $M_{\mathcal{A}}(t)$, if $n_{\mathcal{A}}(0) = 0$